### THE EFFECT OF Cr (VI) ON SILK WORM (BOMBYX MORI) FED ON IN VITRO ACCUMULATED MULBERRY LEAVES

Muhammad Arslan Shoukat<sup>1</sup>, Shahzad Ashraf<sup>2\*</sup>, Muazzam Ali<sup>3</sup>, Zafar Iqbal<sup>4</sup>, Mirza Imran Shahzad<sup>2</sup>, H. R. Chaudary<sup>2</sup>, Nuzhat Sial<sup>3</sup>, Zahra Batool<sup>3</sup>.

1. Department of Agriculture, Yazman, Bahawalpur2. University College of Veterinary and Animal Sciences, TheIslamia University of Bahawalpur, Bahawalpur3. Department of Life Sciences, TheIslamia University of Bahawalpur, Bahawalpur4. Institute of Animal Nutrition and Feed Technology, University of Agriculture, Faisalabad, Pakistan.

### ABSTRACT

Heavy metals take route to higher trophic levels by first of all accumulating in harvestable parts of plants like leaves. Higher the trophic level the amount of metal increase and affect every higher trophic level. Silkworm (*Bombyx mori*) is used as template to assess the biotransformation of heavy metal in a food chain from soil to plant then insect. The mulberry plants were irrigated with synthetic effluent of Cr (VI) (100mg/L with pH 4.5) and the treated soil, mulberry plant leaves, silkworm body, cuticle, alimentary canal, silk glands, silk cocoons and their excreta were sampled to check Cr (VI) in soil and mulberry leaves tend to increase with increase in irrigation times. On the other hand, the contents of Cr (VI) in *B. mori* larvae and the excreta were in considerable amount but decrease with the increase in larval instars, yet most of Cr (VI) remains in its body. Cr (VI) found in *B. mori* body was liable for toxic effects on its life cycle and the body growth and silk production was also inhibited under the effect of Cr (VI) accumulation. A considerable amount of Cr (VI) in larval body increased *B. mori* death rate significantly.

Keywords: *Bombyx mori*, sriculture, bio-transformation, heavy metal.

### **INTRODUCTION**

Silk is one of the most expensive fibers used in textile industry, produced naturally by silk worm (B. mori). Pakistan along with China, India, South Korea and India are the major producers of natural silk. The silk worms not only provide us precious fiber but also has great role in ethhnozoology, as a good remedy for blood pressure, diabetes and other heart diseases (Fernemore and Parkash, 1992). Mulbury leaves are one of the key food sources for silk worms but some of the other plant leave also proved experimentally good as peepal (Nasreen et al., 1989). All over sub-tropical. tropical regions and temperate regions Mulbery tree are grown Mulberry (Morus sp.) is a deciduous, deep-rooted fast growing, perennial tree. Currently, heavy metal contamination is effecting the growth as its root system supports the heavy metal uptake from the soil and deposits it in its leave parts, which are the most preferred source of food for silk worm (Rajaram et al., 2013). Pollution in soil caused by heavy metals is a worldwide problem and is an issue of great environmental concerns (Kabata-Pendies and Pendias, 1992, Kashem and Singh, 1999). As the heavy metals are ingredient of many enzymes and proteins so they are considered essential and crucial for normal development of plant. Toxicity, inhibition of growth and reduce in yield of plants is caused by increase in amount of both essential and non-essential heavy metals in the soil (Costa and Morel, 1993). Use of mineral fertilizers, pesticides, sewage sludge application and anthropogenic source such as the smelting industry, mining, are the main source from where much higher amount is arrived (Michael et al., 2007). Chromium is one of the well-known heavy metal, having significant impacts as proved toxic for agronomic crops (Ghosh and Singh, 2005). The present research is planned to study the phytoremediation of Cr (VI) by mulberry (Morusalba L.) plants and its later effects on silkworm who feed on the mulberry leaves also effect on silk quality produce by silk worm.

The objective of the study is to check the effect

of Cr from synthetic pollutant discharge in

<sup>\*</sup>Corresponding author: e-mail: drshahzadbalouch@yahoo.com

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industrially polluted areas on the growth and productivity of silkworm larvae feeding on mulberry.

### MATERIAL AND METHODS

# Production of Cr (VI) contaminated biomass:

Under the prevailing environmental conditions at Post Graduate Agriculture Research Station which is situated at Jhang Road, Faisalbad, *Morusalba* L. plants were carefully chosen grown in field and irrigated twice a week with serial dilutions of Cr (VI) containing solution ( $K_2Cr_2O_7$  solution with water) with pH maintained at 4.5. Leaves were collected after a pre-determined time of 90 days.

### **Rearing of silkworm:**

The silkworm eggs were obtained by Forest Department from Lahore and these were kept in incubator at 25  $\pm$  1  $^{\circ}$  C and 75  $\pm$  5% R.H. the eggs were hatched within 3 days. The caterpillars which were newly emerged were moved to the rearing trays and were split into two divisions; 1) Treatment fed on grown Cr (VI) contaminated mulberry leaves and 2) was control fraction and was fed by noncontaminated fresh mulberry leaves. In the 5<sup>th</sup> instar of caterpillars some caterpillars were selected. The larvae which left behind were left to pupate. When the larvae are near to pupate the larvae were shifted into pupating chambers made by cardboard. After collection of pupae they are dried in electric oven at 60 ° C for 48 hours, weighted, reeled, and compared with which of control larvae and results were piles up. Later washing with de ionized distilled water (DDW), the treated collected leaves from experimental plants were divided in two fractions, one for feeding of silkworm larvae and second for atomic absorption spectroscopy (AAS) analysis.

### **Dissection of silkworm larvae:**

After the completion of 5<sup>th</sup> instar the larvae were dissected. First the carefully chosen larvae were place into formalin for almost 2 hours and then dissected. The silk glands, alimentary canal and cuticle are obtained after dissection. The obtained body parts after dissection were dried in electric oven for 48 hours at 60°C. after drying the material was exposed to digestion with nitric acid.

# The collection and storage of material for analysis:

Fresh weight of samples was taken and then the samples were dried in an electric oven at 60 ° C for 48 hours. Later the dried material were ground and digested (in Nitric acid and Hydrogen per Oxide), filtered and analyzed. Afterward irrigation of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> solution in water and soil samples were collected fortnightly to detect the deposition of Chromium in the soil. The samples were dried and digested by using method defined by Amacher (1996) as 1.0g of dried soil was taken in Pyrex flask and mixed with 10ml of HNO<sub>3</sub>and left for whole night. Then samples were heated at hot plate at 200°C, after cooling it added 1 mL of HNO<sub>3</sub>and HCLO<sub>4</sub> Heated at 200 °C unless fumes appeared. Sample was cool down and added 1:10 HCl, again heated to 70 °C for one hour. Allowed to cool and made the volume 50 mL by using 1% HCl and filtered and stored for Atomic Absorption Spectrophotometer.

### Analysis of samples and standards:

Javed et al in 2007 used Atomic absorption Spectrophotometer (AAS) Perkin Elmer, Model A Analyst 300 equipped with air acetylene flame, for the analysis of toxic heavy metal i.e. chromium in the samples.

## Atomic absorption spectrophotometer analysis:

Hollow cathode lamps of respective metal ions that provide narrow spectral lines of moderate intensity with air acetylene system were used. Solution containing metal ions was introduced into the flame through aspiration. The metal through thermal dissociation ions were converted into atomic vapors. Atomic vapors were in ground state in which electronic arrangement was stable however part of it absorbed thermal energy emitted from the flame, thus being jumped into excited state in which energy level was higher. When light beam irradiated from hollow cathode lamp, atomic vapors in ground state changes to excited state. The amount of intensity from spectrum to be absorbed was proportional to the number of atoms in the ground state or concentration of metal ions. The results were in the form of spectra between concentration and absorbance (Itanna, 2002).

#### **RESULTS AND DISCUSSION**

This study was conducted to examine the gradual increase of model heavy metal (Cr) in whole the food chain starting from soil to the mulberry and silk worm fed on. The impacts of this accumulation were also investigated by weighing cocoons and determining the mortality rate. The results shown that the accumulation in the soil was increased with the continuous irrigation with Cr (VI) containing solution. After first irrigation the concentration level was  $89.06 \pm 0.02$  and while 60 day regular irrigation the value increased to 221±0.02pm. The heavy metal adhere to the soil particles so consecutive irrigation increased adhered metal ions and remain there due to non-remediation practice. The present continuous irrigation can be parallel to the continuous effluent release from industries containing a bunch of heavy metals. Same was the results were taken by Oklandorf et al.(1986) when studied Zn concentrations in the soil of densely industrial area due to continuous flooding of the area with industrial waste water vs non industrial area. The Cr up taken by mulberry selected to irrigate by Cr (VI) solution. The leaves taken to check the accumulation of up taken metal and analysis revealed same pattern. The number of irrigations and time found directly proportional to the concentration. After 15 days of irrigation the first sample collected the deposition of metal was  $262.96 \pm 0.01$  ppm and the maximum deposition was observed in the last sample of leaves, after 75 days of sampling which was  $425.37 \pm 0.01$  ppm. On the other hand, the metal concentrations in non-treated plants showed minimum accumulation of Cr (VI) in mulberry leaves was studied i.e., 4.9±  $0.01, 6.2 \pm 0.01, 11.71 \pm 0.01, 11.82 \pm 0.01,$  $12.62 \pm 0.01$  ppm after the 15, 30, 45, 60 and 75 days respectively. To feed silkworms, leaves were taken from both treated and non-treated mulberry plants, and accumulation in various parts of the worm was determined. The accumulation of Cr (VI) in the larvae was found to increase with the feeding duration as 16.3  $\pm 0.02$  of the first instar to 69.2  $\pm 0.01$  ppm of the fifth while opposite to it; the larvae fed n non-contaminated leaves declined with every instar. Prince et al., (2001) reported that Cd shows almost similar results regarding accumulation in silkworm. After 5<sup>th</sup> molt, the silk glands developed and the concentration of Cr (VI) was  $289 \pm 0.01$  ppm and  $17 \pm 0.01$  ppm in the larvae fed on contaminated leaves and non- contaminated respectively. As from research of Ali (2010), it shows that with the change in the concentration of Cr (VI) in the effluents the deposition of Cr (VI) in the silk glands of silk worm changes. The minimum deposition was  $4.56 \pm 0.05$  mg/kg and maximum deposition was  $36.67 \pm 0.04 \text{ mg/kg}$ at the 25ppm and 300 ppm effluent concentration respectively. Similarly the deposition of metal in the glands decreases as pH increases as it was maximum. The accumulation of Cr (VI) in silkworm larvae alimentary canal was  $131 \pm 0.01$ . On the other hand, the larvae reared in control conditions the deposition rate found significantly lower as  $9 \pm$ 0.01 ppm. The concentration of Cr(IV) in the excreta also increased with the succeeding molts as the 5th instar had  $193.31 \pm 0.01$  ppm Cr (VI) deposited in faeces while the minimum pass out of metal was found in the 1<sup>st</sup> instar i.e.  $3.91 \pm 0.02$  mg/ kg of Cr (VI). No significant difference found in the concentration Cr in the all the instars fed on non-contaminated leaves as  $10.16 \pm 0.01$ ,  $10.97 \pm 0.01$ ,  $11.01 \pm 0.01$ ,  $11.59 \pm 0.01$ ,  $11.98 \pm 0.01$  ppm during 1st, 2nd, and 3rd, 4th and 5th instar respectively. Many of the previous studies revealed that the concentration of heavy metals in silk worm also depend upon the concentration of metal used in irrigation as at 400 ppm concentration the maximum pass out of heavy metal was observed i.e., 22.3 mg/kg in 5th instar silkworm larvae and the minimum concentration of heavy metal i.e. in synthetic effluent was found at 25 ppm is 9.87. While he was checking the effect of different concentration at different pH levels by Khan (2009). Lisa (1986) and Kazou et al., (1984) reported that by excretion most of the heavy metals were exerted out the larval body. The accumulation of the heavy metal used (Cr) resulted in significant decrease in silk formation evident from the comparative weights of cocoon fed on treated vs control food. Fresh cocoon weights were  $7.2 \pm 0.006$ g and 7.004  $\pm$  0.011 g at 100 mg/L concentration of solution and pH value of 4.5. Whereas the controls were having higher weight of cocoons is  $11.13 \pm 0.004$  and  $11.75 \pm 0.002$  which clearly mean that the silk formation rate was higher in worms fed on non-contaminated leaves. Dry cocoon weights were  $2.6 \pm 0.001$ and 2.7  $\pm$  0.010 for Cr (VI) treatment and for control treatment the dry weights were 3.97  $\pm$ 0.002 and 4.215  $\pm$  0.006. Working with other

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heavy metals, many of the previous investigations revealed the parallel results as Khan (2009) found that Pd accumulation also lowers the production of silk by concentrating the silk glands. Silk worm larvae, s mortality were also tested per 100 larvae in each larval instars and the maximum mortality was found in 1st instar i.e.,  $11 \pm 0.04$  and minimum mortality was found in 5th larval instar i.e.,  $4 \pm$ 0.04. The mortality rate decreased with the succession of molts due to development of resistance and making Cr (VI) as body component, while still the mortality rate in case of treatment stood higher than control ones where the mortality rate remain significantly lower at every instar as maximum in first instar  $(5 \pm 0.05)$  and the successive molts  $3 \pm 0.06$ , 3  $\pm$  0.06, 2  $\pm$  0.05, 2  $\pm$  0.02 during 2nd, 3rd, 4th and 5th instar respectively. Arnandova and Grekoy (2003) also recorded the enhanced mortality rate with increase in accumulation of heavy metals in insect bodies. Along with the metal toxicity, Khan (2009) studied the pH variation effect on mortality rate and found maximum rate at pH 5. The results revealed that the leaves taken from plants grown in polluted areas can significantly decrease production rate and can result in significant loss of this small industry. The plants should grow in control conditions and continuous examination of soil and leaves is necessary to avoid such damages. Soil treatment strategies also require removing this heaping phenomenon of heavy metals. The quality study of silk is still requiring to determine the effect of accumulation on silk produced.

Accum. Of Cr in Insater no/Category		1	2	3	4	5	Cocoon
Accumulation of Cr (VI) in silkworm larvae(ppm)	Treatment	16.3 ±0.02	34.61 ±0.01	47.10 ±0.01	62.0 ±0.02	69.2 ±0.01	Х
Accumulation of Cr (VI) in the feaces of silkworm larvae (ppm)	Control	$8 \pm 0.01$	$4\pm0.01$	$4\pm0.01$	$3\pm0.01$	$3\pm0.01$	Х
	Treatment	$\begin{array}{c} 3.91 \pm \\ 0.02 \end{array}$	$\begin{array}{c} 3.28 \pm \\ 0.01 \end{array}$	$\begin{array}{c} 264.2 \pm \\ 0.01 \end{array}$	$\begin{array}{c} 234.07 \pm \\ 0.02 \end{array}$	$\begin{array}{c} 193.31 \pm \\ 0.01 \end{array}$	Х
	Control	$\begin{array}{c} 10.16 \pm \\ 0.01 \end{array}$	$\begin{array}{c} 10.97 \pm \\ 0.01 \end{array}$	$11.01 \pm 0.01$	11.59 ± 0.01	$\begin{array}{c} 11.98 \pm \\ 0.01 \end{array}$	Х
Accumulation in silk glands(ppm)	Treatment	Х	Х	х	x	$\begin{array}{c} 289 \pm \\ 0.01 \end{array}$	Х
	Control	Х	Х	х	Х	$17 \pm 0.01$	х
Effect of Cr (VI) accumulation on silk yield(gm) Fresh	Treatment	х	х	х	х	Х	$\begin{array}{c} 7.2 \pm \\ 0.006 \end{array}$
	Control	Х	Х	х	Х	Х	$2.6 \pm 0.001$
Dry	Treatment	Х	Х	Х	х	Х	$2.6 \pm 0.001$
	Control	х	Х	х	х	х	$4.215 \pm 0.006$
Effect of Cr (VI) accumulation on the body weight of silk worm larvae(gm)	Treatment	$0.0159 \pm 0.004$	$0.08 \pm 0.005$	$0.69 \pm 0.001$	$4.88 \pm 0.003$	$14.3 \pm 0.002$	
worgin of sink worm in vuo(gin)	Control	$0.021 \pm 0.001$	$0.003 \pm 0.003$	$0.97 \pm 0.006$	$5.42 \pm 0.005$	17.4 ± 0.004	
Effect of Cr (VI) accumulation on the body length of silk worm larvae(cm)	Treatment	$\begin{array}{c} 0.5 \pm \\ 0.004 \end{array}$	$1.09 \pm 0.005$	$2.15 \pm 0.001$	$4.68 \pm 0.003$	5.37 ± 0.002	
	Control	0.49 ± 0.001	1.46 ± 0.003	$2.41 \pm 0.006$	4.83 ± 0.005	5.71 ± 0.004	
Mortality rate of silk worm larvae(n)	Treatment	$11\pm0.02$	$9\pm0.04$	$8\pm0.01$	$6\pm0.03$	$4 \pm 0.04$	
	Control	$5 \pm 0.05$	$3 \pm 0.06$	$3 \pm 0.06$	$2 \pm 0.05$	$2 \pm 0.02$	

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