THREE DIMENSIONAL QUANTITATIVE STRUCTURE ACTIVITY RELATIONSHIPS OF 3-[3-(N-4-FLUROBENZYL-N-2 PYRIDYLAMINO) PROPYLAMINO]-4-ETHYLAMINO-1, 2, 5- THIAZOLE-1-OXIDE (HISTAMINE H₁ RECEPTOR ANTAGONIST) AS AN ANTI-ULCER AND ANTI-ALLERGY

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ABSTRACT

This computer aided drug design based conformational and potential energy quantitative structure activity relationship established to find out best fit conformations with minimum potential energy, active structural and molecular sites for strong binding of 3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5- thiazole-1-oxide ($C_{19}H_{23}FN_6OS$) as Histamine H_1 receptor antagonist which is previously derived from Cimetidine and extracted by leaves of *Espinheria santa*. Potential energy calculated for lower limit by Kitagorodasky potential energy function and in-silico approaches. This compound crystalline in monoclinic system, with unit cell dimensions $a = 6.686 \text{ A}^\circ$, $b = 14.717 \text{ A}^\circ$ and $c = 20.850 \text{ A}^\circ$, $\beta = 97.83^\circ$. Present work find out new bond pair (N_{21} - C_{15}) with strong bonding site and suggests that 3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5- thiazole-1-oxide has shown minimum potential energy -0.00000983 Kcal/mole as Histamine H_1 receptor antagonist at $\omega_1 = 280^\circ$ and $\omega_2 = 340^\circ$ (Figure-9). It would supports stable and strong binding of antihistaminic compound to their receptors in allowed active region. The compound ($C_{19}H_{23}FN_6OS$) in this active conformation would be helpful for the treatment of allergy and different kind of ulcers such as stomach and duodenum ulcers.

Key words: $(C_{19}H_{23}FN_6OS)$, Histamine H_1 receptor antagonist, Allergy, Ulcer, Coordinate, Monoclinic and rectangular unit cell system, Receptor, Conformational and quantitative analysis.

INTRODUCTION

Drug discovery leads to the concept of computer aided drug design (Florence *et al.*, 2002; James *et al.*, 2004; Michael Williams, 2004 and Tommy Liljeforl, 2004). It is use to identify drugs, drug-like compounds and assist clinical testing (Weerasak Samee *et al.*, 2004). It is dependent on the flexibility, lowest energy conformers of the most rigid compounds and superimpose. This may use to develop new compounds of minimum energy conformer, desired positions functional groups and most favorable binding mode within active sites of receptor (Munikumar Reddy Doddareddy *et al.*, 2004 and Gupta, 1989).

Histamine (2, 4-imidazolyl ethylamine) (Fig. 1) is a natural, dibasic, vasoactive amine; neuromodulator and neurotransmitter that stimulate stomach cells to produce acid (Abbas *et al.*, 1994 and Bannister *et al.*, 1994). It is located in most of body tissues and gastrointestinal tract (Abbas *et al.*, 1994; Bach, Jean-Francois, 1982; Froese, 1980 and Gallin, *et al.*, 1980). This single polypeptide chain produced by histamine decarboxylation and stored in effectors cell (Goldie, 1990; Jawetz Melnick *et al.*, 2002; Metcalfe *et al.*, 1997 and Monroe *et al.*, 1997). Release of histamine causes hay fever, bronchial infection, initial phase of asthma, (Bach, Jean-Francois, 1982 and Paul, William, 1984), urticaria (Rang *et al.*, 1995), gastrointestinal tract, suffocation and massive degranulation of mast cells (Abbas *et al.*, 1994, Bach, 1982, Sompayrac, Lauran, 1999, Spiegelberg, 1989). Its deficiency leads to lower levels of melatonin and decreased fat metabolism in the brain that can cause stress (Sompayrac, Lauran, 1999). It is used to treat different kinds of ulcers and allergies (Abbas *et al.*, 1994 and Bach, Jean-Francois, 1982).

This paper describes computer aided drug design based conformational and potential energy quantitative structure activity relationship analysis of 3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5- thiazole-1-oxide, $C_{19}H_{23}FN_6OS$ (Bannister *et al.*, 1994). It is histamine H_1 receptor antagonist, (Fig. 2), previously derived from cimetidine (histamine H_2 receptor, Fig. 3) (Bannister *et al.*, 1994 and Patrícia Maria Ferreira *et al.*, 2004), extracted by *Maytenus aquifolium* and *Maytenus ilicifolia* (*Espinheria santa*) leaves (Fig. 4) (Patrícia Maria Ferreira *et al.*, 2004). The data has been taken from literature (Bannister *et al.*, 1994). This current research also correlates and compares with previous study (Bannister *et al.*, 1994) for determination of minimum potential energy, bond lengths, bond angles, new molecular and structural sites by using potential energy function and *in-sillico* approaches. The $C_{19}H_{23}FN_6OS$ contains a heterocyclic head linked by a four-atom chain to a dipolar tail (Bannister *e.t al.*, 1994), (Fig. 3). It has anti-allergy and anti-ulcer properties (Abbas *et al.*, 1994; Bach,

Jean-Francois.,1982; Bannister *et al.*, 1994; Froese, 1980; Gallin *et al.*, 1980; Sompayrac, Lauran, 1999 and Spiegelberg, 1989). It can suppresses the histamines-induced wheal (swelling), flare (vasodilatation) (Abbas *et al.*, 1994; Bach, Jean-Francois., 1982; Sompayrac, Lauran, 1999 and Spiegelberg, 1989), motion sickness (Abbas, *et al.*, 1994; Goldie, 1990; Jawetz Melnick *et al.*, 2002, and Metcalfe *et al.*, 1997), certain types of headaches, Cohn's disease, acute multiple sclerosis and some stomach secretary conditions (Abbas *et al.*, 1994; Bach, Jean-Francois., 1982 and Sompayrac, Lauran, 1999), response by blocking the binding of histamine to its receptors. It is available as over-the-counter drugs, such as, Azatadine (Optimine), carbinoxamine/pseudoephedrine (cardec), cetrizine (zyrtec), cyproheptadine (periactin), fexofenadine (allegra) and loratadine (Bach, Jean-Francois, 1982; Monroe *et al.*, 1997 and Paul William, 1984). The search for histamine receptor antagonists by structural modification has lead to discovery of several new antihistaminic for clinical use (Abbas *et al.*, 1994; Bach, Jean-Francois, 1982; Gallin *et. al.*, 1980; Jawetz Melnick *et al.*, 2002; Sompayrac, Lauran, 1999 and Spiegelberg, 1989).

METHODS

This work is not only describes the conformational analysis, (Lingling Shen., 2003 and Manuel *et. al.*, 1992) potential energies determination and optimization of (3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5- thiazole-1-oxide) but also helpful to design potent candidate for medicinal use. It's based on quantitative predictions and action mechanism of a drugs (Munikumar *et al.*, 2004; Gupta, 1989; Tommy Liljeforl., 2004 and; Weerasak Samee *et al.*, 2004).

The structure of (3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5- thiazole-1-oxide) was constructed on gaussian-98 (Gaussian 98, Gaussian, Inc., Pittsburgh PA, 2003 USA). Structural modification, minimization and charges were calculated by sybyl 6.9 (SYBYL 6.9. Tripose Inc., 1699). Molden and Kitaigorodskii potential energy function (Kitaigorodskii, 1961 and Ramakrishan et al., 1965) ware used to optimizations and calculate potential energy respectively. The bond lengths, bond angles, rotation of coordinates and potential energy ware calculated by in-sillico approaches GW-Basic 3.0, Statistica 0.5. Gaussian 98 and Molden. In order to determine allowed conformation, the contact distances between the atoms in adjacent residues have to be examined by using the criteria for the minimum Vander Waals contact distance. X, Y, Z Cartesian coordinates on X, Y, Z-axis collected from literature (Bannister et al., 1994). The fractional coordinates X_f , Y_f , Z_f ware multiplied by Unit Cell dimensions a = 6.686 A°, b = 14.717 A° and c = 20.850 A°, converted into monoclinic and rectangular system by formula such as,

$$X_r = Y_m + Z_m \cos \beta (97.83^\circ)$$

$$Y_r = Y_m$$

$$Z_r = Z_m \sin \beta (97.83^\circ)$$

Where

 $X_{m,}$ Y_{m} and Z_{m} (mono-clinic coordinates) on X, Y and Z axis. $X_{r,}$ $Y_{r,}$ and Z_{r} (rectangular coordinates) on X, Y and Z axis,

Contact distance between atoms can be calculated by the following relationship,

$$r_{ij} = -(X_2 - X_1)^2 + (Y_2 - Y_1)^2 + (Z_2 - Z_1)^2$$

 X_1 , Y_1 , Z_1 (coordinates of first atom in molecule)

 X_2 , Y_2 , Z_2 (coordinates of second atom in molecule)

 r_{ij} = distance between non-bonded interaction

Bond angles can be calculated by given below formula,

$$Cos\theta = - \frac{L_3 L_1 L_2}{2 L_1 * L_2}$$

L = Bond length

Coordinates after rotation:

$$\begin{split} L &= X_1 \text{-} X_2 / l \\ M &= Y_1 \text{-} Y_2 / l \\ N &= Z_1 \text{-} Z_2 / l \end{split}$$

L, M, N (the direction of cosines on the axis of rotation with respect to the chosen system of coordinates).

```
a = \cos \omega/2
    b = L \sin \omega/2
    c = M \sin \omega/2
    d = N \sin \omega/2
Where.
\omega (Angle of rotation 0° to 360°)
a, b, c, d (Variables)
X' = (a^2 + b^2 - c^2 - d^2) X + 2 (bc - ad) Y + 2 (bd + ac)Z
Y' = 2 (bc + ad) X + (a^2 - b^2 + c^2 - d^2) Y + 2 (cd - ab) Z

Z' = 2 (bd - ac) X + 2 (cd + ad) Y + (a^2 - b^2 - c^2 + d^2) Z
(X', Y' Z' are the coordinates after rotation)
Potential energies can be calculated by potential energy function (Kitaigorodskii, 1961 and Ramakrishan et al.,
1965).
V = 3.5 [8600 e^{-13z} - .04 / z^6]
Z = r_{ii}/r_{\cdot}
r_{\cdot} = equilibrium distance
r_{ii} = distance between non-bonded interaction
Z = variable
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RESULTS AND DISCUSSION

In present work, atomic coordinates were converted into fractional, monoclinic and rectangular coordinate systems by gaussion-98, Molden and GW-Basic (Tables 1, 2 and 3). The bond lengths (Table 4) and bond angles (Table 6) were calculated newly by gaussion-98 and molden and correlated with previously calculated study bond lengths (Table 5) and bond angles (Table 7) where no above-mentioned computational approaches were used. The C_{18} , N_{19} , C_{8} , C_{9} , N_{21} - C_{15} , S_{22} - C_{14} , O_{23} - C_{15} , C_{20} - C_{10} , O_{24} - C_{12} , O_{26} - C_{12} , O_{27} - O_{13} , O_{28} - O_{13} , O_{21} - O_{15} , O_{22} - O_{14} , O_{23} - O_{15} , O_{20} - O_{10} , O_{24} - O_{12} , O_{26} - O_{12} , O_{27} - O_{13} , O_{28} - O_{15} , O_{29} - O_{29}

These results shown that 3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5-thiazole-1-oxide (Histamine H_1 receptor antagonist) has shown maximum potential energy (10.1 Kcal/mole at ω_1 = 60° and ω_2 = 140°) and minimum potential energy (-0.00000983 Kcal/mole at ω_1 = 280° and ω_2 = 340°) for pair N_{21} - C_{15} (Fig. 16). The positive marked area shows non-allowed or disfavored region where $C_{19}H_{23}FN_6OS$ would not interact with its receptor because of rigidity, short contacts and clops between atoms. While negative marked area shows allowed or favored region (no short contacts and clops found) where $C_{19}H_{23}FN_6OS$ would show its antiallergy and anti-ulcer properties by interact or bind strongly with its specific receptors. These kinds of analysis are beneficial for discovery of new antihistamines and would helpful for the treatment of allergy, different kind of ulcer such as stomach and duodenum ulcers and medicinal search.

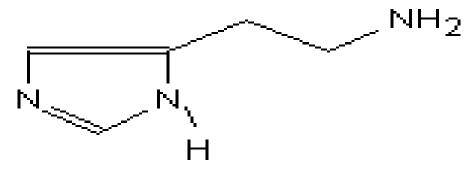
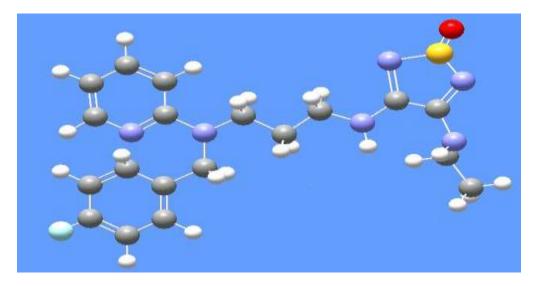


Fig.1. Histamine (2, 4-imidazolyl ethylamine).



 $Fig. 2.\ Structure\ of\ 3-[3-(N-4-Flurobenzyl-N-2\ pyridylamino)\ propylamino]-4-ethylamino-1,\ 2,\ 5-\ thiazole-1-oxide\ (C_{19}H_{23}FN_6OS)\ by\ Gaussian.$

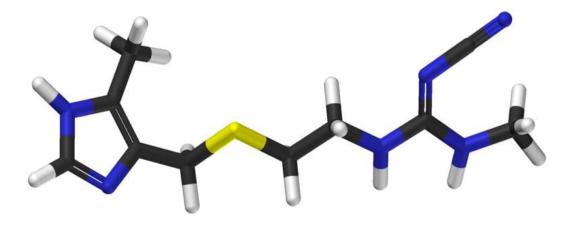


Fig.3. Structure of Cimetidine (Histamine H_2 receptor).



Fig.4. Espinheria santa leaves.

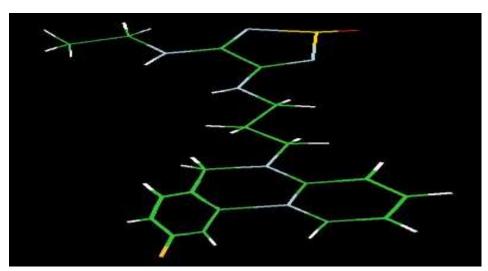


Fig.5. stereo view of $C_{19}H_{23}FN_6OS$ (before optimization) through Molden.

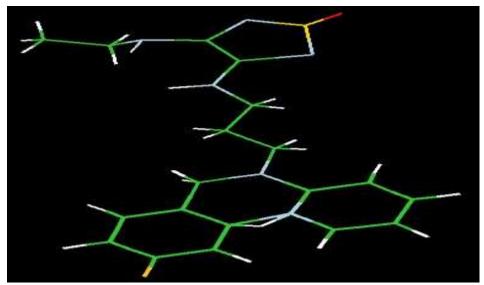


Fig.6. stereo view of $C_{19}H_{23}FN_6OS$ (after optimization) through Molden.

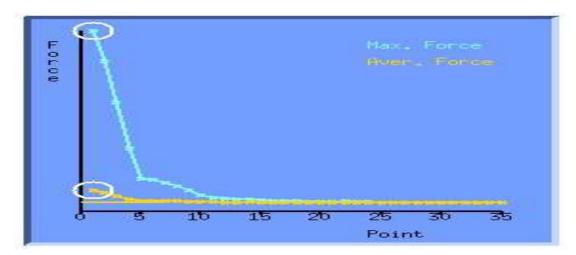


Fig.7. Compound C₁₉H₂₃FN₆OS shown Maximum force (cyne) and average force (yellow) at initial stage.

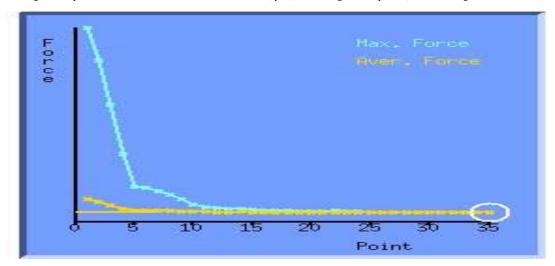


Fig.8. Compound C₁₉H₂₃FN₆OS shown maximum force (cyne) and average force (yellow) at end stage.

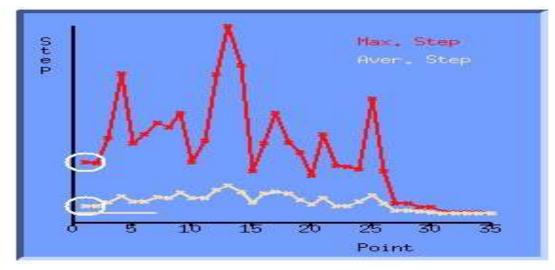


Fig.9. Compound C₁₉H₂₃FN₆OS shown maximum steps (red) and average steps (yellow) involved in energy minimization at initial stage.

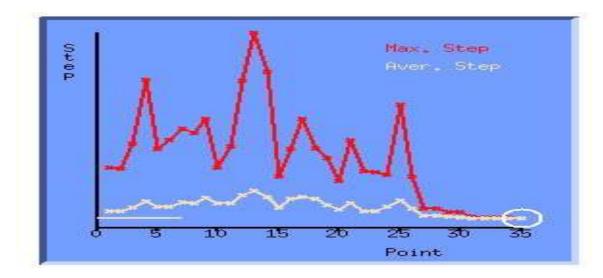


Fig. 10. Compound $C_{19}H_{23}FN_6OS$ shown maximum steps (red) and average steps (yellow) involved in energy minimization at end stage.

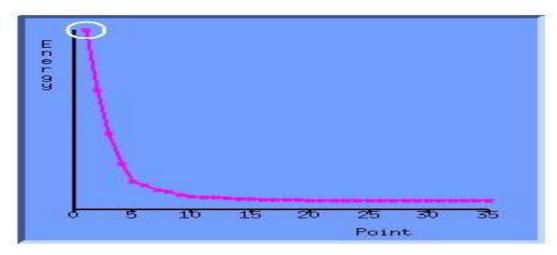


Fig.11.Compound C₁₉H₂₃FN₆OS shown maximum energy (purple) at initial stage.

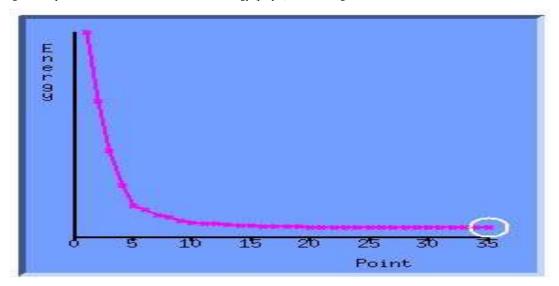


Fig.12. Compound $C_{19}H_{23}FN_6OS$ shown minimum energy (purple) at end stage.

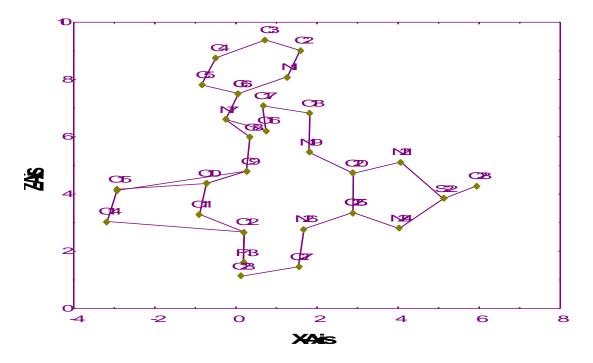


Fig-13. 010 projection of Histamine H1 receptor antagonist (3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5-thiazole-1-oxide) on ZX- axis.

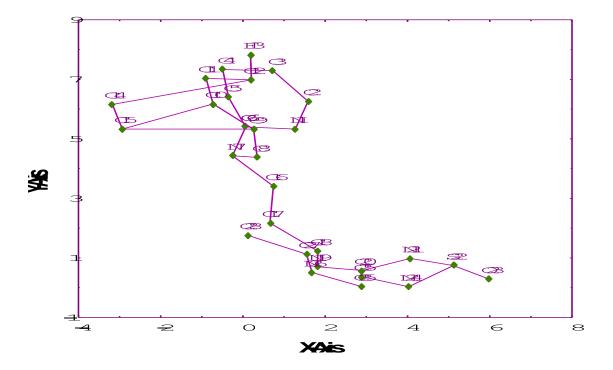


Fig-14. 001 projection of Histamine H1 receptor antagonist (3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5-thiazole-1-oxide) on YX- axis.

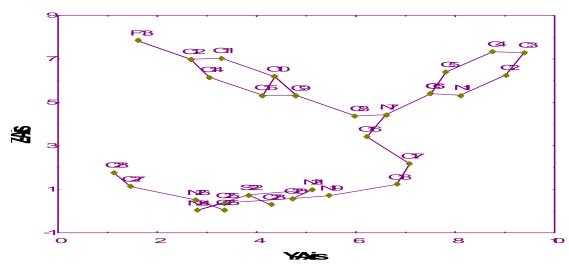


Fig-15. 100 projection of Histamine H1 receptor antagonist (3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino]-4-ethylamino-1, 2, 5-thiazole-1-oxide) on ZY- axis.

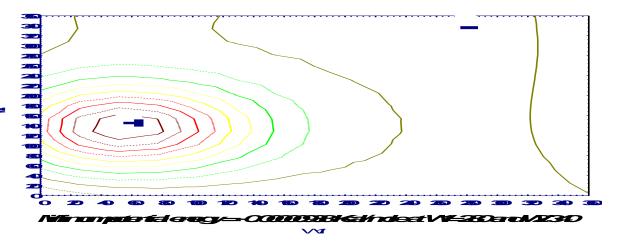


Fig-16. Potential energy calculated for pair N_{21} - C_{15} ; minimum potential energy (-0.00000938 Kcal/mole at ω_1 = 280° and ω_2 = 340°) and maximum potential energy (10.1 Kcal/mole at ω_1 = 60° and ω_2 = 140°) were found.

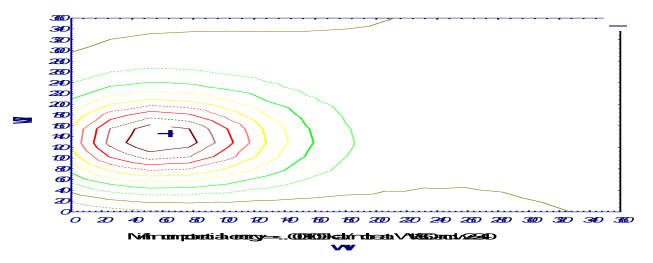


Fig-17. Potential energy calculated for pair S_{22} -- C_{14} ; minimum potential energy (-0. 0001009 Kcal /mole at ω_1 = 360° and ω_2 = 340°) and maximum potential energy (87.3 Kcal /mole at ω_1 = 60° and ω_2 = 140°) were found respectively.

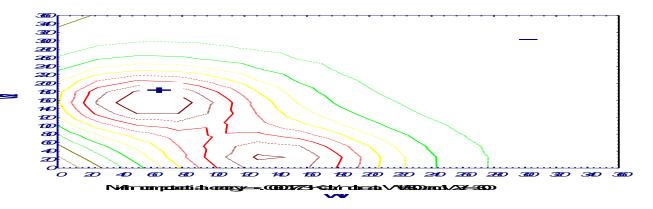


Fig-18. Potential energy calculated for pair O_{23} -- C_{15} ; minimum potential energy ((-0. 00001773 Kcal/mole at ω_1 = 300° and ω_2 = 300°) and maximum potential energy (408.3 Kcal/mole at ω_1 = 60° and ω_2 = 160°) were found respectively.

Table 1. Fractional coordinates of 3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino] 4-ethylamino-1, 2, 5- thiazole-1-oxide, $C_{19}H_{23}FN_6OS$.

Serial	Fractional Coordinates On X-axis		Fractional Coordinates On Y-axis	Fractional Coordinates On Z-axis
1	N ₁	0.3001	0.5504	0.2580
2	C ₂	0.3655	0.6127	0.3032
3	C ₃	0.2592	0.6376	0.3528
4	C ₄	0.0770	0.5951	0.3548
5	C ₅	0.0042	0.5309	0.3095
6	C 6	0.1210	0.5103	0.2615
7	N ₇	0.0530	0.4486	0.2146
8	C ₈	0.1419	0.4070	0.2118
9	C 9	0.1484	0.3263	0.2576
10	C ₁₀	0.0184	0.2969	0.2990
11	C ₁₁	0.0109	0.0109 0.2237	
12	C ₁₂	0.1746	0.1816	0.3380
13	F ₁₃	-0.1884	0.1091	0.3792
14	C ₁₄	-0.3481	0.2067	0.2981
15	C ₁₅	-0.3290	0.2804	0.2580
16	C ₁₆	0.1805	0.4229	0.1653
17	C ₁₇	0.1431	0.4812	0.1052
18	C ₁₈	0.2986	0.4636	0.0593
19	N ₁₉	0.2836	0.3711	0.0342
20	C ₂₀	0.4453	0.3210	0.0279
21	N ₂₁	0.6288	0.3479	0.0477
22	S ₂₂	0.7803	0.2604	0.0366
23	O ₂₃	0.9024	0.2913	0.0138
24	N ₂₄	0.6087	0.1899	0.0015
25	C ₂₅	0.4325	0.2280	0.0018
26	N ₂₆	0.2622	0.1891	0.0247
27	C ₂₇	0.3540	0.0992	0.0545
28	C ₂₈	0.053	0.0763	0.0853

Table 2. Monoclinic coordinates of 3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino] 4-ethylamino-1, 2, 5- thiazole-1-oxide, $C_{19}H_{23}FN_6OS$.

Serial	Atoms	Monoclinic Coordinates On X-axis	Monoclinic Coordinates On Y-axis	Monoclinic Coordinates On Z-axis
1	N ₁	2.600	8.100	5.379
2	C ₂	2.443	9.017	6.321
3	C ₃	1.733	9.383	7.355
4	C ₄	0.514	8.758	7.397
5	C 5	0.028	7.813	6.453
6	C ₆	0.809	7.510	5.452
7	N ₇	0.354	6.602	4.474
8	C ₈	0.948	5.989	4.416
9	C 9	0.992	4.802	5.370
10	C ₁₀	0.123	4.369	6.234
11	C ₁₁	0.072	3.292	7.099
12	C ₁₂	1.167	2.672	7.047
13	F ₁₃	-1.259	1.605	7.906
14	C ₁₄	-2.327	3.042	6.215
15	C ₁₅	-2.199	4.126	5.379
16	C ₁₆	1.206	6.223	3.446
17	C ₁₇	0.956	5.461	2.193
18	C ₁₈	1.996	7.081	1.236
19	N 19	1.896	5.461	0.713
20	C ₂₀	2.977	4.724	0.581
21	N ₂₁	4.204	5.120	0.994
22	S ₂₂	5.217	3.832	0.763
23	O ₂₃	6.033	4.287	0.287
24	N ₂₄	4.06	2.794	0.031
25	C ₂₅	2.89	3.355	0.037
26	N ₂₆	1.753	2.782	0.514
27	C ₂₇	1.698	1.459	1.136
28	C ₂₈	0.354	1.122	1.778

Table 3. Rectangular coordinates of 3-[3-(N-4-Flurobenzyl-N-2 pyridylamino) propylamino] 4-ethylamino-1, 2, 5- thiazole-1-oxide, $C_{19}H_{23}FN_6OS$.

Serial	Atoms	Rectangular Coordinates On X-axis	Rectangular Coordinates On Y-axis	Rectangular Coordinates On Z-axis
1	N ₁	1.273	8.1	5.329
2	C ₂	1.582	9.017	6.262
3	C ₃	0.73	9.383	7.287
4	C4	-0.492	8.758	7.328
5	C ₅	-0.851	7.813	6.392
6	C ₆	0.066	7.51	5.401
7	N ₇	-0.255	6.602	4.432
8	C ₈	0.347	5.989	4.374
9	C ₉	0.26	4.802	5.32
10	C ₁₀	-0.072	4.369	6.176
				Cont'd

11	C ₁₁	-0.894	3.292	7.033
12	C ₁₂	0.207	2.672	6.981
13	F ₁₃	0.182	1.605	7.632
14	C ₁₄	-3.174	3.042	6.157
15	C ₁₅	-2.932	4.126	5.329
16	C ₁₆	0.737	6.223	3.414
17	C ₁₇	0.657	7.081	2.172
18	C ₁₈	1.827	6.822	1.224
19	N 19	1.799	5.461	0.706
20	C ₂₀	2.898	4.724	0.576
21	N ₂₁	4.068	5.12	5.114
22	S ₂₂	5.114	3.832	0.743
23	O ₂₃	5.994	4.287	0.285
24	N ₂₄	4.065	2.782	0.03
25	C ₂₅	2.886	3.355	0.037
26	N ₂₆	1.683	2.782	0.51
27	C ₂₇	1.543	1.459	1.125
28	C ₂₈	0.112	1.122	1.761

Tables 4-5. Calculated bonds length.

Tables 6-7. Calculation for bond angles.

Present work

Previous work

Present work

Previous work

Atomic Pairs	Calculated Bonds Leng th (A°)	Atomic Pairs	Calculated Bonds Leng th (A°)	Atomic Bonds Angles	Calculated Bonds Angles (°)	Atomic Bonds Angles	Calculated Bonds Angl es (°)
N ₁ C ₂	1.343	N ₁ C ₂	1.345	C ₂ - N ₁ - C ₆	117.8	C ₈₋ N ₇₋ C ₆	121.7
N ₁ C ₆	1.345	N ₁ C ₆	1.346	$N_1 - C_2 - C_3$	123.6	C ₁₆ - N ₇ - C ₆	120.2
C ₂ C ₃	1.382	C ₂ C ₃	1.381	C ₂ - C ₃ - C ₄	116.8	C ₁₆ - N ₇ - C ₈	118.1
C ₃ C ₄	1.374	C ₃ C ₄	1.375	C ₃₋ C ₄₋ C ₅	121.5	C ₉₋ C ₈₋ N ₇	114.2
C ₄ C ₅	1.382	C ₄ C ₅	1.377	C ₄₋ C ₅₋ C ₆	117.4	C ₁₇₋ C ₁₆₋ N ₇	112.8
C ₅ C ₆	1.382	C ₅ C ₆	1.385	N ₁₋ C ₆₋ C ₅	122.7	$C_{18-}C_{17-}C_{16}$	111.9
C ₆ N ₇	1.365	C ₆ N ₇	1.367	N ₁₋ C ₆₋ N ₇	117.6	C ₁₇₋ C ₁₈₋ N ₁₉	111.3
N ₇ C ₈	0.861	N ₇ C ₈	1.434	C ₅₋ C ₆₋ N ₇	119.6	S ₂₂₋ N ₂₁₋ C ₂₀	106.5
N ₇ C ₁₆	1.471	N ₇ C ₁₆	1.472	C ₆₋ N ₇₋ C ₈	110.8	N ₂₁₋ S ₂₂₋ O ₂₃	105.3
C ₈ C ₉	1.521	C ₈ C ₉	1.528	C ₆₋ N ₇₋ C ₁₆	120.2	N ₂₁₋ S ₂₂₋ N ₂₄	98.90
C ₈	1.063	C ₈	1.384	C _{8 -} N ₇₋ C ₁₆	225.4	N ₂₄ - S ₂₂ - O ₂₃	105.2
C ₉ C ₁₀	1.375	C ₉ C ₁₀	1.385	N _{7 -} C ₈₋ N ₉	118.3	S ₂₂ - N ₂₄ C ₂₅	107.3
C ₁₀ C ₁₁	1.387	C ₉ C ₁₅	1.388	N _{7 -} C ₈₋ C ₁₆	99.29	N ₂₆ - C ₂₅ - N ₂₄	123.8
C ₁₁ C ₁₂	1.265	C ₁₀ C ₁₁	1.381	C ₉₋ C ₈₋ C ₁₆	139.0	C ₂₀₋ C ₂₅₋ C ₂₄	112.9
C_{14}	1.386	C_{10}	1.382		132.3	C ₁₈₋ N ₁₉₋ C ₂₀	122.4
				C ₈₋ C ₉₋ C ₁₀		N ₁₉₋ C ₂₀₋ N ₂₁	122.9
C ₁₆ C ₁₇	1.511	C ₁₄ C ₁₅	1.386	C ₉₋ C ₁₀₋ C ₁₁	135.6	C ₂₅₋ C ₂₀₋ N ₂₁	114.0
C_{17} C_{18}	1.528	C_{16} C_{17}	1.509	$C_{10-} C_{11-} C_{12}$	104.4	C ₂₅₋ C ₂₀₋ N ₁₉	123.0

C ₁₈ N ₁₉	1.456	C ₁₇ C ₁₈	1.529	N ₇₋ C ₁₆₋ C ₈	215.2	C ₂₀₋ C ₂₅₋ N ₂₆	123.3
N ₁₉ C ₂₀	1.330	C ₁₈ N ₁₉	1.458	N ₇₋ C ₁₆₋ C ₁₇	112.8	C_{27-} N_{26-} C_{25}	122.2
C ₂₀ N ₂₁	1.301	N ₁₉ C ₂₀	1.330	C ₈₋ C ₁₆₋ C ₁₇	148.0	$N_{26-}C_{27-}C_{28}$	112.2
C ₂₀ C ₂₅	1.470	C ₂₀ N ₂₁	1.301	C ₁₆₋ C ₁₇₋ C ₁₈	111.9		
S 22O23	1.097	C ₂₀	1.500	C ₁₇₋ C ₁₈₋ N ₁₉	111.3		
N ₂₄ C ₂₅	1.305	S 22O23	1.487	C ₁₈₋ N ₁₉₋ C ₂₀	122.4		
C ₂₅ N ₂₆	1.414	N ₂₄ C ₂₅	1.304	N ₁₉₋ C ₂₀₋ N ₂₁	122.9		
N ₂₆ C ₂₇	1.466	C ₂₅ N ₂₆	1.306	N ₁₉₋ C ₂₀₋ C ₂₅	123.0		
		N ₂₆	1.459	N ₂₁₋ C ₂₀₋ C ₂₅	113.8		
				C ₂₀₋ C ₂₅₋ N ₂₄	113.2		
				C ₂₀₋ C ₂₅₋ N ₂₆	105.1		
				N ₂₄₋ C ₂₅₋ C ₂₆	126.6		
				C ₂₅₋ C ₂₆₋ C ₂₇	125.9		

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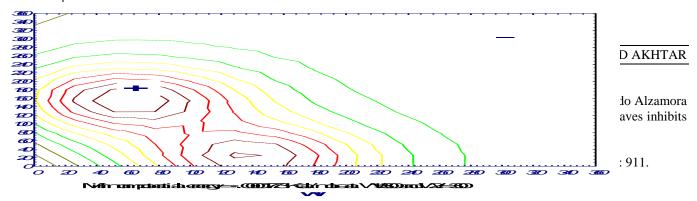
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(Accepted for publication October 2008)