POTASSIUM REMOVAL FROM THREE ALLUVIAL SOILS BY GREENHOUSE AND LABORA TOR Y PROCEDURES

S.M. Mehdi, I.A. Chaudhry, A.M. Ranjha and M. Qadir Department of Soil Science, University of Agriculture, Falsalabad,

A pot experiment was conducted on the Shahdara, Nabipur and LyaUpur soil series to study the relationship between potassium uptake by wheat cv. LU26S and the amount extracted by various extractants, N~40Ac, 0.5N MgOAc, 0.025N CaCl like ~ 20.1 N and IN HNO)':1 Nitrogen at the rate of 0, 150 and 200 kg N ha and pjlosphorus at the rate of 0, 44 and 66 kg P ha were applied. The highest correlation (r = 0.984) between IN NH₄0Ac extractable potassium and plant uptake was obtained when no NP fertilizers were applied while 0.025 N CaCl gave the highest correlations of r = 0.936 and r = 0',815 for $N150^P44$ and $N_{200}P66$ treatments,

INTRODUCTION

There have been many attempts to characterize nutrient supplying power of soils by simulating the feeding action of plant roots and through determining the available fraction of nutrients during the vegetative growth (Soltanpour and Schwab, 1977). There is a general consensus that most of the cultivated soils in Pakistan have sufficient supply of available potassium for plant growth because of the dominance of Elite glay mineral (Ranjha, 1988). However, under intensive cropping, many soils are likely to be exhausted with respect to their potassium reserves and thus are becoming responsive to potassium fertilization (Malik et al., 1987).

In Pakistan, soils are designated as low, medium or high on the basis of NH_40Ac test but without correlating the test

values with crop response. The soil testing laboratories need simple, rapid, and cheap but reasonably accurate methods for potassium analysis. We present data regarding the correlation between the amount of K extracted with different extractants and K uptake by wheat cv. LU26S.

MATERIALS AND METHODS

The soil samptes were collected from 0-15 cm and 15-45 cm depths of the three soil series viz., Shahdara, Nabipur and Lyallpur (Table 1). These were air dried, ground, passed through a 2mm sieve and added to plastic pots (3.5 kg capacity) at the rate of 3 kg per pot. The fertilizer combinations were NOPO (cont.roi), N150P44 and N200P66 Nit:ogen as urea a~d phosphorus as slngle superptiosphate were applied before sowing the crop. Six grains of wheat variety LU26S were sown in each pot and on germination thinned to three plants per pot. Canal water was used for irrigation. The crop was harvested at maturity. Grain and straw yields were recorded (data not given). potassium concentration in both the parts of the plant was determined (U.S. Salinity Laboratory Staff, 1954). Potassium uptake by wheat from each pot was calculated by using the formulae:

K uptake
$$(mg \text{ pot-}^{-1})$$
 by grain $=$ $\frac{\% \text{ K in grain } \times \text{ Grain } \text{ yield } (g)}{100} \times 1000$

K uptake $(rng \text{ pot-}^{-1})$ by straw $=$ $\frac{\% \text{K in straw } \times \text{ Straw } \text{ yield } (g)}{100} \times 1000$

K uptake by wheat $(rng \text{ pot-}^{-1})$ $=$ $(rng \text{ wheat } \text{ K})$ uptake by $(rng \text{ pot-}^{-1})$ $=$ $(rng \text{ wheat } \text{ K})$ uptake by $(rng \text{ pot-}^{-1})$ $=$ $(rng \text{ wheat } \text{ has } \text$

To characterize the soils used, evaluation of total nitrogen and available phosphorus was done by Jackson (1962) and Watanabe and Olsen (1965), respectively. Extraction of potassium was done by H 0, IN NH $_4$ 0Ac, 0.5N MgOAc, 0.025N CaC12 and IN HN0 $_3$ ·

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RESUL TS AND DISCUSSION

Potassium uptake by wheat

Potassium uptake by wheat increased with the-application of nitrogen and phosphorus to aJI the soil series (Table 2). This indicated that ample available potassium was present for improving the Nand P fertilizer use efficiency. Potassium uptake by the crop was more from surface soil (0-15 cm) than from subsurface soil 05-45 cm) except for the Shahdara soil series On the control) and the Nabipur soil series in aJI the treatments. It may be due to the fact that soil samples of the Nabipur soil series were taken from the paddy area where most of the potassium might have leached from surface to subsurface soil.

Potassium extracted by various extractants

The water extractable potassium from all the soils was lower compared to that with the other extractants (Table 3). The extract ants 1N NH₄0Ac, 0.5N MgOAc and, 0.025N CaCl extracted greater amount of potassium, partiaJIy exchangable fraction also. The higher potassium extracted by IN NH 0Ac than that with 0.5N MgOAc and 0.025N CaCl might be due to the size of NH; being similar to that of potassium. Even NH; can extract potassium from interlayer sites of 2:1 clay minerals while calcium and magnesium cannot because of larger ionie size. Potassium extracted by IN HN0₃ from all the soil series was higher than all the other extractants. The acid might have extracted non-exchangeable potassium in addition to available fraction.

Correlation between potassium uptake by wheat and potassium extracted by various extractants

Potassium extracted by all the extractants was positively correlated with potassium uptake by wheat in all the treatments. IN NH₄0Ac-K was highly correlated (r = 0.984) with potassium uptake by wheat plants in the control, 0.025N CaCl₂ proved best extractant in T7 (N150P44) and T3 (N₂₀₀P66) giving values of r = 0.936 and r = "0.81", respectively. Similar values calculated for H₂0 and IN HN0₃ were nonsignificant for all the treatments excepf in the control for 1N HNO₅. This leads to the conclusion that all the extractants other than 11_2 0 and IN HN0₃ can provide good index of soil available potassium for optimum growth of wheat.

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REFERENCES

- Jackson, M.L. 1962. Soil Chemical Analysis. Constable and Co. Ltd., London. 496 pp.
- Malik, D.M., R.A. Chaudhry and G. Hassan. 1987. Crop response to K application in the Punjab. Proc, Workshop on "Role of Potassium in Improving Fertilizer Use Efficiency". NFDC/PARC, Islamabad, Pakistan. pp.1-33.
- Ranjha, A.M. 1988. Morphological, mineralogical and chemical properties of some soils of Pakistan. Ph.D. Thesis, Dept. of Soil Sci., Univ, Agri., Faisalabad,
- Soltanpour, P.N. and A.P. Schwab. 1977. A new soil test for simultaneous extraction of macro and micronutrients in alkali soils. Commun. Soil Sci. Plant Anal. 8: 195-207.
- U.S. Salinity Laboratory Staff. 1954. Diagnosis and Improvement of Saline and Alkali Soils. U.S.D.A. Handbook No.60, U.S. Govt. Printing Office, Washington, D.C. 160pp.
- Watanabe, F.S. and S.R. Olsen. 1965. Test of an ascorbic acid method for determining phosphorus in water and NaHCO extracts from soil. Soil Sci, Soc. Amer. Proc, 29: 677-67~