

EFFICIENCY OF GYPSIUM GRADES AND QUALITY OF LEACHING
WATER FOR RECLAIMING A SALINE-SODIC SOIL.
n, CHEMICAL IMPROVEMENT OF SOIL

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Gypsum grades (5-16, 16-25, 25-60, 60-100, > 100 mesh) @ 100 % GR of 15 cm soil column and four synthetic waters (EC 0.6 + SAR 6; EC 1.0 + SAR 12; EC 2.0 + SAR 18; EC 4.0 dS m⁻¹ + SAR 30), in all possible combinations, were tested for reclaiming a loamy clay saline-sodic soil in a laboratory experiment. The results indicated that the dissolution of gypsum grades varied only by less than 1 me l⁻¹ in all the synthetic waters. The reduction in EC and SAR increased as the gypsum fineness increased and/or brackishness of water decreased. The pH was decreased by coarser grades more than that with the finer ones, whereas the increasing brackishness of waters caused progressively lower decrease in pH

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INTRODUCTION

The method for reclamation of salt-affected soils depends upon the Source and nature of soluble as well as exchangeable ions; physical, chemical and mineralogical properties; presence of lime or gypsum in soil, cost and the availability of both the Ca-source and irrigation water.

Being cheaper and easily available, gypsum is preferred for soil reclamation. The fineness to which gypsum must be ground is a matter of economic consideration, though the finer grades may improve a soil earlier. Low solubility of gypsum, i.e. 25-30 me l⁻¹ (Bresler ~ al. 1982) discourages the use of

coarse particles while the excess of sodium in sodic soils favours the use of coarse grades (Keren & Shainberg, 1981).

The low hydraulic conductivity of a sodic/saline-sodic soil can be maintained or increased, irrespective of sodium saturation, by using sufficiently concentrated salt-solutions (Muhammed et al., 1969). Since most of the ground waters in the Punjab (75 % of the existing discharge of wells) are saline, saline-sodic or sodic in nature (Malik et al., 1984), it was planned to explore the possibilities to utilize these waters for leaching during reclamation of a saline-sodic soil being treated with different gypsum grades.

MATERIALS AND METHODS

A bulk sample from surface 30 cm layer of t1-1 Gandhra soil series (loamy clay in texture, EC = 14 dS m⁻¹, SAR = 59, pH = 9.1) was collected, sun-dried, ground and passed through a 2 mm sieve. A 30 cm high columns each with 2400 g soil were prepared in 72 PVC Pipes (45 x 8.75 cm) with a uniform bulk density of 1.33 Mg m⁻³. Gypsum @ 100 % GR per 15 cm layer was mixed with the surface 15 cm before processing the pipes for uniform bulk density. This was achieved by vertically dropping the pipes four times from a height of 5 cm. The pipes were leached with 90 cm of synthetic waters under continuous submergence at 26 ± 2°C. The pipes were arranged in completely randomized design with three replications at a uniform height from floor. After the termination of the experiment, each soil column was divided into two equal depths, i.e., 0-15 and 15-30 cm. These samples were analysed by the methods of the U.S. Salinity Lab. Staff (1954).

Size Limit Of Gypsum Particles

GO = No gypsum	G ₁ = 5-16 mesh;
G ₂ = 16-25 mesh ;	G ₃ = 25-60 mesh;
G ₄ = 60-100 mesh;	G ₅ = > 100 mesh.

Synthetic Waters

Different synthetic waters were prepared by using NaCl,

Na₂S₂O₈·j, CaCl₂·2H₂O and MgSO₄·7H₂O salts where Ca:Mg::4: 1, and cr:sgfl...: r:r ratios were rr~Yljoined alongwith the following EC (dS m⁻¹) and SAR (mmol l⁻¹) levels.

$$\begin{aligned} W_1 &= EC \ 0.6 + SAR \ 6; & W_2 &= EC \ 1.0 + SAR \ 12; \\ W_3 &= EC \ 2.0 + SAR \ 18; & W_4 &= EC \ 4.0 + SAR \ 30. \end{aligned}$$

RESULTS AND DISCUSSION

The dissolution of gypsum grades was determined by shaking a known weight (5g) of each grade in 100 ml of each synthetic water on a wrist action shaker for 30 minutes at 20°C. The concentration of Ca + Mg in filtrate was recorded to be 26.43, 26.27, 26.15 and 26.06 me l⁻¹, respectively in W₁, W₂, W₃ and W₄ while that for the gypsum grades G₁, G₂, G₃ and G₄ was, respectively 25.94, 26.09, 26.21, 26.36 and 26.58 me l⁻¹. The differences due to synthetic water composition as well as due to size of gypsum particles were significant, though the variation in Ca + Mg concentration was less than 1 me l⁻¹.

Soil EC_e

A maximum significant decrease in EC occurred with W₁, followed by W₂, W₃ and W₄ (Table 1). Relatively higher decrease with W₁, W₂ and W₃ though safe for most of agricultural crops (Mass, 1971), may be due to progressive higher EC + SAR of waters used for leaching. All the waters decreased EC more from the upper soil layers than from the lower ones, the efficiency of waters remained of the same order as above. These differences appear to be due to higher water potential being useful to dissolve and carry the salts downward more from the upper than the lower layers. As the water passed through the soil, it became loaded with salts, resulting in decreased carrying power and hence less removal of salts.

The gypsum grades as well as the leaching alone (control) lowered the EC to less than 4 dS m⁻¹ except the G₁ where it was 4.16 dS m⁻¹. However, the treatments differed statistically and, in general, the finer gypsum grades decreased the EC more than the coarser ones. Similar results were reported by

gypsum particles having range of size distribution (16-25 mesh) when saline-sodic water is available, can be used successfully for timely reclamation of native soils, where a variety of agricultural crops can be grown.

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